

# 1. Functional group

- From Wikipedia, the free encyclopedia



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
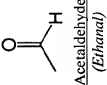

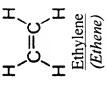

For other uses, see [Functional group \(disambiguation\)](#).

In organic chemistry, **functional groups** are specific groups of atoms within molecules, that are responsible for the characteristic chemical reactions of those molecules. The same functional group will undergo the same or similar chemical reaction(s) regardless of the size of the molecule it is a part of.









The following is a list of common functional groups. In the formulas, the symbols R and R' usually denotes an attached hydrogen, or a hydrocarbon side chain of any length, but may sometimes refer to any group of atoms. Below is an image of multiple functional groups found in organic chemistry.













(For convenience, the basic functional groups covered in General Biology are also listed [here](#))




<u>Chemical class</u>	Group	Formula	Graphical Formula	Prefix	Suffix	Example
<u>Acyl halide</u>	Haloformyl	RCOX		haloformyl-	-oyl halide	 Acetyl chloride ( <i>Ethanoyl chloride</i> )

<u>Alcohol</u>	<u>Hydroxyl</u>	ROH	$\text{R}-\text{O}-\text{H}$	hydroxy-	-ol	 Methanol
<u>Aldehyde</u>	<u>Aldehyde</u>	RCHO	 <u>Acetaldehyde</u> (Ethanal)	oxo-	-al	 Methane
<u>Alkane</u>	<u>Alkyl</u>	RH	 Ethylene (Ethene)	alkenyl-	-ane	
<u>Alkene</u>	<u>Alkenyl</u>	$\text{R}_2\text{C}=\text{CR}_2$		alkenyl-	-ene	

<u>Alkyne</u>	<u>Alkynyl</u>	$RC \equiv CR'$	$R \equiv R'$	alkynyl-	-yne	$H-C \equiv C-H$ <u>Acetylene</u> ( <i>Ethyne</i> )
<u>Amide</u>	<u>Carboxamide</u>	$RCONR_2$	$\begin{array}{c} R' \\   \\ R-C=O \\   \\ R \end{array}$	carboxamido-	-amide	$\begin{array}{c} NH_2 \\   \\ R-C=O \end{array}$ <u>Acetamide</u> ( <i>Ethanamide</i> )
<u>Amines</u>	<u>Primary amine</u>	$RNH_2$	$\begin{array}{c} H \\   \\ R-N-H \\   \\ R \end{array}$	amino-	-amine	$\begin{array}{c} H \\   \\ H-C-N-H \\   \quad   \\ H \quad H \end{array}$ <u>Methylamine</u> ( <i>Methanamine</i> )
	<u>Secondary amine</u>	$R_2NH$	$\begin{array}{c} H \\   \\ R-N-R' \end{array}$	amino-	-amine	$\begin{array}{c} CH_3 \\   \\ H-N-CH_3 \end{array}$ <u>Dimethylamine</u>
	<u>Tertiary amine</u>	$R_3N$	$\begin{array}{c} R'' \\   \\ R-N-R' \end{array}$	amino-	-amine	$\begin{array}{c} \diagup \\ :N \\ \diagdown \end{array}$ <u>Trimethylamine</u>

	<u>4° ammonium ion</u>	$R_4N^+$	$\begin{array}{c} R_4 \\   \\ N^+ \\ / \backslash \\ R_1 \ R_2 \end{array}$	ammonio-	-ammonium	 Choline
<u>Azo compound</u>	<u>Azo (Diimide)</u>	$RN_2R'$	$\begin{array}{c} R \\   \\ N=N \\   \\ R' \end{array}$	azo-	-diazene	 Methyl orange
<u>Toluene derivative</u>	<u>Benzyl</u>	$RCH_2C_6H_5$ RBn		benzyl-	1- ( <i>substituent</i> )toluene	 Benzyl bromide (1-Bromotoluene)
<u>Carbonate</u>	<u>Carbonate ester</u>	ROCOOR			alkyl carbonate	
<u>Carboxylate</u>	<u>Carboxylate</u>	$RCOO^-$	 	carboxy-	-oate	 Sodium acetate (Sodium ethanoate)

<u>Carboxylic acid</u>	<u>Carboxyl</u>	RCOOH		carboxy-	-oic acid	 Acetic acid (Ethanoic acid)
<u>Cyanates</u>	<u>Cyanate</u>	ROCN		cyanato-	alkyl cyanate	
	<u>Thiocyanate</u>	RSCN		thiocyanato-	alkyl thiocyanate	
<u>Ether</u>	<u>Ether</u>	ROR'		alkoxy-	alkyl alkyl ether	 Diethyl ether (Ethoxyethane)
<u>Ester</u>	<u>Ester</u>	RCOOR'			-oate	 Ethyl butyrate (Ethyl butanoate)
<u>Haloalkane</u>	<u>Halo</u>	RX	$R-X$	halo-	alkyl halide	 Chloroethane (Ethyl chloride)
<u>Hydroperoxide</u>	<u>Hydroperoxy</u>	ROOH		hydroperoxy-	alkyl hydroperoxide	 Methyl ethyl ketone peroxide
<u>Imine</u>	<u>Primary ketimine</u>	RC(=NH)R'		imino-	-imine	

	<u>Secondary ketimine</u>	$RC(=NR)R'$	$\begin{array}{c} R'' \\ \diagup \\ N \\ \diagdown \\ R' \end{array}$	imino-	-imine	
	<u>Primary aldimine</u>	$RC(=NH)H$	$\begin{array}{c} H \\ \diagup \\ N \\ \diagdown \\ R \end{array}$	imino-	-imine	
	<u>Secondary aldimine</u>	$RC(=NR')H$	$\begin{array}{c} R' \\ \diagup \\ N \\ \diagdown \\ R \end{array}$	imino-	-imine	
<u>Isocyanide</u>	<u>Isocyanide</u>	RNC	$R-\overset{+}{N}\equiv C^{-}$	isocyano-	alkyl isocyanide	
<u>Isocyanates</u>	<u>Isocyanate</u>	RNCO	$R-\overset{+}{N}\equiv C=O$	isocyanato-	alkyl isocyanate	
	<u>Isothiocyanate</u>	RNCS	$R-\overset{+}{N}\equiv C=S$	isothiocyanato-	alkyl isothiocyanate	 Allyl isothiocyanate
<u>Ketone</u>	<u>Ketone</u>	RCOR'	$\begin{array}{c} R^2 \\ \diagup \\ O \\ \diagdown \\ R^1 \end{array}$	keto-, oxo-	-one	 Methyl ethyl ketone (Butanone)
<u>Nitrile</u>	<u>Nitrile</u>	RCN	$R-C\equiv N$	cyano-	alkanenitrile alkyl cyanide	 Benzonitrile (Phenyl cyanide)